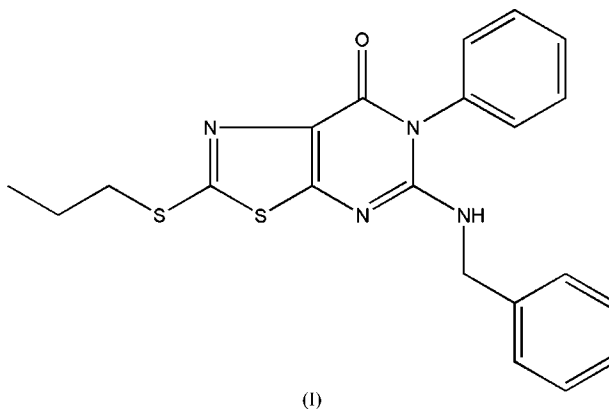


Nian-Yu Huang,^a Ju-Zhen Yuan,^a
Rui-Jun Xu,^b Ming-Wu Ding^{a*}
and Hong-Wu He^a^aKey Laboratory of Pesticide and Chemical
Biology of Ministry of Education, College of
Chemistry, Central China Normal University,
Wuhan 430079, People's Republic of China,
and ^bDepartment of Chemistry, Central China
Normal University, Wuhan 430079, People's
Republic of ChinaCorrespondence e-mail:
ding5229@yahoo.com.cn

Key indicators

Single-crystal X-ray study
 $T = 292\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.007\text{ \AA}$
Disorder in main residue
 R factor = 0.066
 wR factor = 0.183
Data-to-parameter ratio = 13.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.5-Benzylamino-6-phenyl-2-propylsulfanyl-
thiazolo[5,4-*d*]pyrimidin-7(6*H*)-oneIn the title compound, $\text{C}_{21}\text{H}_{20}\text{N}_4\text{OS}_2$, the two rings of the
fused thiazolo[5,4-*d*]pyrimidine system are almost coplanar. In
the crystal structure, hydrogen-bonded tetramers are formed
via $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds.Received 9 November 2005
Accepted 30 November 2005
Online 7 December 2005

Comment

Thiazolo[5,4-*d*]pyrimidines are purine analogue derivatives
and they have potentially useful biological properties (El-
Bayouki & Basyouni, 1988). In recent years, we have been
engaged in the synthesis of derivatives of various heterocycles
via the aza-Wittig reaction (Ding, Yang & Zhu, 2004; Ding,
Chen & Huang, 2004). The title compound, (I), may be used as
a new precursor for obtaining bioactive molecules and we
report here the X-ray crystallographic analysis of this
compound.The molecular structure of the title compound is shown in
Fig. 1, and selected bond distances and angles are given in
Table 1. All the ring atoms in the thiazolo[5,4-*d*]pyrimidine
system are essentially coplanar. The *n*-propylthio group is
disordered [occupancies 0.583 (6)/0.417 (6); unprimed/primed
atoms; Fig. 1].In the crystal structure of (I), symmetry-related molecules
are linked *via* $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds to form tetramers.
Details are given in Table 2 and Fig. 2.

Experimental

Compound (I) was prepared by adding benzylamine (0.22 g) to a
solution of ethyl 5-[(phenylimino)methyleneamino]-2-(propylthio)-
thiazole-4-carboxylate (2 mmol) in dry dichloromethane (10 ml). The
solution was stirred for 2 h at 298 K. The solvent was then removed
and anhydrous ethanol (10 ml), containing several drops of EtONa in
EtOH, was added. The mixture was stirred for 6 h at room
temperature. The solution was then concentrated under reduced

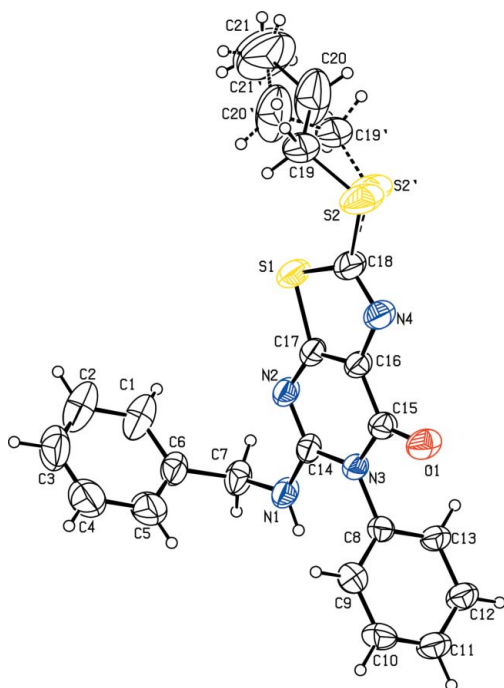


Figure 1

View of the molecular structure of (I), showing the atom labelling scheme and with displacement ellipsoids drawn at the 50% probability level. Both disorder components are shown.

pressure and the residue was recrystallized from ethanol to give compound (I), in a yield of 85% (m.p. 460 K). Suitable crystals were obtained by vapour diffusion of ethanol and dichloromethane at room temperature. $^1\text{H NMR}$ (CDCl_3 , 400 MHz): 7.17–7.58 (*m*, 10H, Ph–H), 4.57–4.59 (*m*, 3H, N1–H, C7–H), 3.26–3.3(*t*, 2H, C19–H), 1.78–1.82 (*m*, 2H, C20–H), 1.03–1.06 (*t*, 3H, C21–H).

Crystal data

$\text{C}_{21}\text{H}_{20}\text{N}_4\text{OS}_2$
 $M_r = 408.53$
 Tetragonal, $P4_2/n$
 $a = 23.3140$ (6) Å
 $c = 7.7925$ (4) Å
 $V = 4235.6$ (3) Å³
 $Z = 8$
 $D_x = 1.281$ Mg m⁻³

Mo $K\alpha$ radiation
 Cell parameters from 2882 reflections
 $\theta = 2.5$ – 16.9°
 $\mu = 0.27$ mm⁻¹
 $T = 292$ (2) K
 Plate, colourless
 $0.30 \times 0.20 \times 0.06$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: none
 29639 measured reflections
 3737 independent reflections

2138 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.080$
 $\theta_{\text{max}} = 25.0^\circ$
 $h = -27 \rightarrow 27$
 $k = -27 \rightarrow 27$
 $l = -9 \rightarrow 9$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.066$
 $wR(F^2) = 0.183$
 $S = 1.04$
 3737 reflections
 268 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0722P)^2 + 1.9388P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.29$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.30$ e Å⁻³

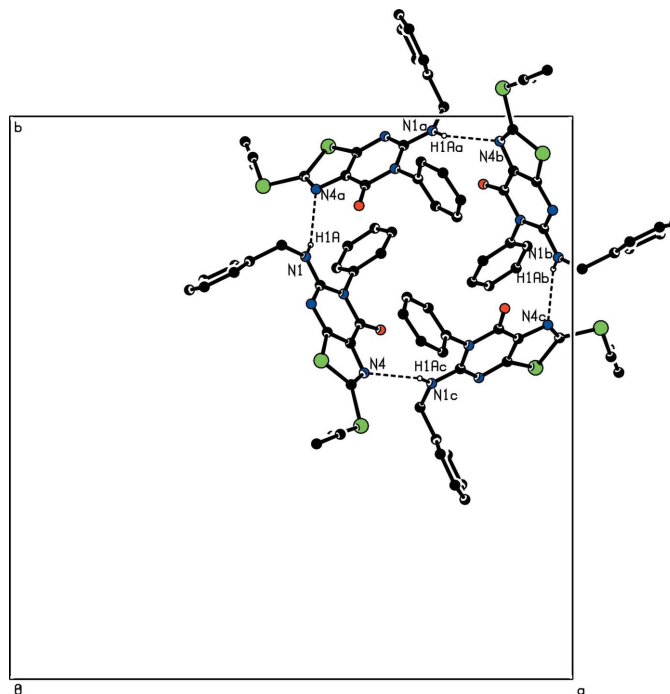


Figure 2

A partial view of the crystal packing of (I), showing the formation of an N–H...N hydrogen-bonded tetramer. Hydrogen bonds are shown as dashed lines and atoms labelled a, b and c correspond to symmetry operations $(y, \frac{3}{2} - x, \frac{3}{2} - z)$, $(\frac{3}{2} - x, \frac{3}{2} - y, z)$ and $(\frac{3}{2} - y, x, \frac{3}{2} - z)$, respectively. H atoms not involved in hydrogen bonds have been omitted.

Table 1

Selected geometric parameters (Å, °).

S1–C17	1.739 (4)	N1–C14	1.345 (4)
S1–C18	1.758 (5)	N2–C17	1.353 (5)
S2–C18	1.798 (8)	N2–C14	1.309 (5)
S2–C19	1.776 (12)	N3–C15	1.413 (5)
S2'–C19'	1.714 (17)	N3–C8	1.447 (4)
S2'–C18	1.740 (11)	N3–C14	1.394 (5)
O1–C15	1.223 (5)	N4–C16	1.387 (5)
N1–C7	1.442 (5)	N4–C18	1.281 (6)
C17–S1–C18	88.2 (2)	O1–C15–N3	119.9 (3)
C18–S2–C19	102.8 (5)	O1–C15–C16	127.6 (3)
C18–S2'–C19'	115.0 (8)	N3–C15–C16	112.5 (3)
C7–N1–C14	121.1 (3)	N4–C16–C15	123.7 (3)
C14–N2–C17	113.6 (3)	N4–C16–C17	116.7 (3)
C8–N3–C14	120.3 (3)	S1–C17–C16	109.4 (3)
C8–N3–C15	116.9 (3)	N2–C17–C16	127.7 (4)
C14–N3–C15	122.8 (3)	S1–C17–N2	122.9 (3)
C16–N4–C18	109.9 (3)	S1–C18–S2'	117.0 (5)
N1–C7–C6	114.2 (3)	S1–C18–N4	115.9 (3)
N3–C8–C9	121.3 (3)	S2'–C18–N4	125.9 (5)
N3–C8–C13	118.6 (3)	S1–C18–S2	127.3 (4)
N2–C14–N3	123.8 (3)	S2–C18–N4	116.0 (4)
N1–C14–N3	116.9 (3)	S2–C19–C20	121.2 (8)
N1–C14–N2	119.3 (3)	S2'–C19'–C20'	112.7 (13)

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1–H1A...N4 ⁱ	0.86	2.38	3.069 (4)	138

Symmetry code: (i) $y, -x + \frac{3}{2}, -z + \frac{3}{2}$.

The H atoms were placed in calculated positions and treated as riding atoms, with C–H = 0.93–0.97 Å, N–H = 0.86 Å, and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent C or N atom})$ and $1.5U_{\text{eq}}(\text{methyl C atom})$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINTE* (Bruker, 2000); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

References

- Bruker (2000). *SMART* (Version 5.618) and *SAINTE* (Version 6.10). Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2001). *SHELXTL*. Version 5.0. Bruker AXS Inc., Madison, Wisconsin, USA.
- Ding, M. W., Chen, Y. F. & Huang, N. Y. (2004). *Eur. J. Org. Chem.* **18**, 3872–3878.
- Ding, M. W., Yang, S. J. & Zhu, J. (2004). *Synthesis*, **1**, 75–79.
- El-Bayouki, Kh. A. M. & Basyouni, W. M. (1988). *Bull. Chem. Soc. Jpn*, **61**, 3794–3796.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.